

**AMENDMENTS TO THE SPECIFICATION**

**Please replace the last paragraph bridging pages 3 and 4 with the following rewritten paragraph:**

The metal-polishing composition as described in Japanese Patent Application Laid-Open (Kokai) No. ~~8-837808-83980~~ containing benzotriazole effectively provides a flat surface and prevents dishing. However, polishing rate detrimentally decreases due to strong anti-corrosion effect of benzotriazole. The polishing composition as described in Japanese Patent Application Laid-Open (Kokai) No. 9-55363 containing 2-quinolinecarboxylic acid is not preferred for industrial use, since 2-quinolinecarboxylic acid is a remarkably expensive material.

**Please replace the sixth paragraph on page 7 with the following rewritten paragraph:**

[14] The polishing composition as recited in [12] or [13] above, wherein the surfactant is at least one species selected from the group consisting of an alkylaromatic-sulfonic acid or a salt thereof, polyoxyethylene alkyl phosphoric acid or a salt thereof, alkyl phosphoric acid or a salt thereof, and a fatty acid or a salt thereof.

**Please replace the seventh paragraph on page 7 with the following rewritten paragraph:**

[15] The polishing composition as recited in any one of [12] to [14] above, wherein the content of the surfactant is in a range of 5 mass% or less.

**Please replace the first paragraph on page 8 with the following rewritten**

**paragraph:**

[18] The polishing composition as recited in any one of [1] to [17] above, wherein the protective-film-forming agent comprises at least one species selected from the group consisting of benzotriazole, tolyltriazole, hydroxybenzotriazole, carboxybenzotriazole, benzimidazole, tetrazole, and quinaldinic acid.

**Please replace the second paragraph on page 8 with the following rewritten**

**paragraph:**

[19] The polishing composition as recited in [18] ~~or [19]~~ above, wherein the content of the protective-film-forming agent is in a range of 10 mass % or less.

**Please replace the fifth paragraph on page 8 with the following rewritten**

**paragraph:**

[22] The polishing composition as recited in [20] or [21] above, wherein the content of the alkali substance is in a range of 10 mass% or less.

**Please replace the eighth paragraph on page 8 with the following rewritten**

**paragraph:**

[25] The polishing composition as recited in any one of [1] to [24]~~[23]~~ above, wherein the content of the abrasive is in a range of 30 mass% or less.

**Please replace the last paragraph bridging pages 8 and 9 with the following rewritten paragraph:**

[28] The polishing composition as recited in any one of [1] to ~~[27]~~[26] above, wherein a ratio ( $P_{RR}/B_{RR}$ ), between a metal film polishing rate ( $P_{RR}$ ) for polishing a metal film formed on a substrate having trenches such that the metal film fills the trenches, or polishing a metal film formed on a substrate having trenches and a barrier metal film formed on the substrate such that the metal film fills the trenches, and a metal film polishing rate ( $B_{RR}$ ) for polishing a flat blanket metal film, is 3.5 or more.

**Please replace the third full paragraph on page 9 with the following rewritten paragraph:**

[31] A polishing method comprising forming a metal film provided on the substrate having trenches such that the metal film fills the trenches, by use of the polishing composition as set forth in any one of [1] to ~~[28]~~[27] above.

**Please replace the fourth full paragraph on page 9 with the following rewritten paragraph:**

[32] A polishing method comprising forming a barrier metal film on a substrate having trenches, and polishing, by use of the polishing composition as recited in any one of [1] to ~~[28]~~[27] above, a metal film provided on the substrate such that the metal film fills the trenches.

**Please replace the fifth full paragraph on page 9 with the following rewritten paragraph:**

[33] A polishing method comprising a metal film, wherein a metal film formed on a substrate having trenches such that the metal film fills the trenches, or a metal film formed on a substrate having trenches and a barrier metal film formed on the substrate such that the metal

film fills the trenches, has protrusions, and corners of the protrusions are preferentially polished by the composition as ~~set~~<sup>set</sup> forth in any one of [1] to ~~[28]~~<sup>[27]</sup> above.

**Please replace the third full paragraph on page 10 with the following rewritten paragraph:**

[37] The method for producing a substrate, the method comprising a step of polishing, through the polishing method as recited in any one of [31] to ~~[33]~~<sup>[36]</sup> above, a metal film provided on a substrate having trenches such that the metal film fills the trenches.

**Please replace the first full paragraph on page 29 with the following rewritten paragraph:**

Molecular weight of each of the synthesized compounds was determined through gel permeation chromatography (GPC) (reduced to polyethylene glycol). In the present invention, a commercial product was also employed, and molecular weight of the commercial product was also determined. The employed commercial product was VPI55K18P (hereinafter abbreviated as 18P) (1-vinylimidazole-1-vinylpyrrolidone (1:1) copolymer, product of BASF) and VPI55K72W (hereinafter abbreviated as ~~72W~~<sup>18P</sup>, product of BASF).

**Please replace the paragraph on page 42 with the following rewritten paragraph:**

The results are shown in Table 6. The rate of polishing tantalum barrier film decreased and erosion was more prevented, as increase in amount of azole ~~18PP~~<sup>18</sup>. However, since addition of ~~18PP~~<sup>18</sup> in an excessive amount was prone to increase copper polishing rate, step reduction and dishing prevention effect tended to be impaired to degrees slightly higher than

those of Comparative Example 3 employing no 18PP18. Therefore, the amount of 18PP18 is required to be appropriately modified to attain well balance in the composition.

**Please replace the last paragraph bridging pages 50 and 51 with the following rewritten paragraph:**

DBS stands for dodecyl benzene sulphate, BTA benzotriazole, and THFA ~~tetrahydrofurfuryl~~amine. ~~The~~ colloidal silica used has a particle size of 70-80nm.

**Please replace Table 14 on page 56 with the following new Table 14:**

Table 14

Ex. Com.Ex.	Polishing Pressure	<u>Cu</u> <u>(blanket)</u> <u>polishing</u> <u>rate B<sub>RR</sub></u>	<u>Cu</u> <u>(pattern)</u> <u>polishing</u> <u>rate P<sub>RR</sub></u>	P <sub>RR</sub> /B <sub>RR</sub>	Step reduction	Dishing
Ex.32	14 kPa	130 nm/min	550 nm/min	4.23	AA	50-60nm
Ex.33	10 kPa	110 nm/min	450 nm/min	4.09	AA	40-60nm
Ex.34	20 kPa	140 nm/min	630 nm/min	4.50	AA	70-80nm